Acids and Bases



Acid				Base		
Strength of Acid	Ka	Name	Formula	Formula	Name	Strength of Base
Strongest	Large	Perchloric acid	HCIO.	CIO,-	Perchlorate ion	Weakest
1	3.2×10*	Hydroiodic acid		1	Todide	
	1.0×10*	Hydrobromic acid	HBr	Bre	Bromide	T 72
	1.3×10^{4}	Hydrochloric acid	HCI	CI-	Chloride	
	1.0×100	Sulfuric acid	H _i SO ₄	HSO,-	Hydrogen sulfate ion	
	2.4×10 ¹	Mitric acid	HNO ₂	NO _y -	Nitrate ion	
		Hydronium ion	H ₂ O+	нуо	Water	
100	5.4×10 ⁻³	Oxelic acid	но,с,о,н	HO ₂ C ₂ O ₂ -	Hydrogen exalate ion	
	1.3×10-	Sulfurous acid	H _a SO _a	HSO,-	Hydrogen sulfite ion	
	1.0×104	Hydrogen sulfate ion	HSO: -	50.3	Sulfate ion	-
	7.1×10^{-3}	Phosphoric acid	н,РО.	H ₂ PO ₂ -	Dihydrogen phosphate ion	
	7.2×10 ⁻⁴	Nitrous acid	HNO _a	NO ₂ -	Nitrite ion	
	6.6×10*	Hydrofluoric acid	HF	F-	Fluoride ion	
	1.8×10-	Methanoic acid	нооун	HCO ₂ -	Methanoate ion	
	6.3×10 ⁻⁶	Benzoic acid	С"Нускон	C,H,000-	Benzoate ion	
Medium	5.4×10°	Hydrogen oxalate ion	HO ₂ C ₂ O ₂ -	O ₂ C ₂ O ₂ 2-	Oxalate ion	Medium
	1.8×10 ⁻⁵	Ethanoic acid	СН,СООН	CH/COO	Ethanoate (acetate) ion	
	46.48 ± 1.0^{-7}	Carbonic acid	н, со,	HCO ₃ -	Hydrogen carbonate ion	
	1.1×10^{-r}	Hydrosulfuric acid	H ₁ S	HS-	Hydrogen sulfide ion	
	6.3×10°	Dihydrogen phosphate ion	H ₂ PO ₄ +	HPO _x 2-	Hydrogen phosphate ion	
	6.2×10*	Hydrogen suifite ion	HS-	50,1	Sulfite ion	
	2.9×10^{-6}	Hypochlorous acid	HCIO	CIO-	Hypochlorite ion	
	6.2×10-4	Hydrocyanic acid	HCN	CN-	Cyanide ion	
	5.8×10 ⁺⁺	Ammonium ion	NHL +	NH ₃	Ammonia	
	5.8×10 **	Boric acid	H,80,	H ₂ BO ²	Dihydrogen borate	
	4.7×10^{-11}	Hydrogen carbonate ion	HCO ₂ -	CO2F	Carbonate ion	
	4.2×10 ⁻¹³	Hydrogen phosphate ion	HPO, P	PO. P	Phosphate Ion	
	1.8×10-45	Dihydrogen borate ion	н,воз-	нво, -	Hydrogen borate ion	
	$1.3\!\times\! 10^{-13}$	Hydrogen sulfide ion	HS-	S.1-	Sulfide ion	1
	1.6×10**	Hydrogen borate ion	нво, -	BO, *	Borate ion	
Weekest		water	H ₂ O	OH-	Hydroxide	Stronges



Lab 16 1 Creating Acid Base Strength Table

Society of Chemical Industry (Great Britain)

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Reviewing Lab 16 1 Creating Acid Base Strength Table: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "Lab 16 1 Creating Acid Base Strength Table," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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