

**Section 2.1 – continued**

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	_____
_____	_____
_____	_____
_____	_____
_____	_____

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 18. It appears in the periodic table.
- _____ 19. It has no line periodic table.
- _____ 20. Group 1 elements.
- _____ 21. Elements that are strong oxidizing agents.
- _____ 22. Group 8 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most elements have lines in the periodic table for metals.
- _____ 25. Group 1 elements except for hydrogen are metals in the solid state.
- _____ 26. Group 2 elements are the alkaline earth metals.
- _____ 27. Group 16 elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 18 elements are inert nonmetallic elements known as noble gases.
- _____ 29. Elements have properties of both metals and nonmetals.

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IM Harris



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