

# THE PARTICLE THEORY

The particle theory describes the behaviour of matter. According to this theory, all matter is made up of tiny particles, which are constantly in motion. The behaviour of these particles is influenced by the temperature and pressure of their surroundings.



Fill in the blanks using the word bank provided. Next, draw a diagram to show how particles are arranged in a solid, liquid and gas.

## WORD BANK

packed      movement      shape      fit      vibrate      slide      closely  
freely      energetic      contact      electrical      independently

### SOLID



In a solid, the particles are tightly \_\_\_\_\_ together and \_\_\_\_\_ in place. They have a fixed \_\_\_\_\_ and volume because the particles are held closely together by strong forces of attraction. The particles can only move in a fixed position and cannot move past one another.

### LIQUID



In a liquid, the particles are still close together but have more freedom of \_\_\_\_\_ than in a solid. They can \_\_\_\_\_ past one another, allowing liquids to take the shape of their container. However, they still have a fixed volume because the particles are still \_\_\_\_\_ packed.

# Light Particle Model Quiz 3 Answers

**David Halliday, Robert Resnick, Jearl Walker**

### **Light Particle Model Quiz 3 Answers:**

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