

U8 Rev WS pg 2.jpg



Chemistry I Chapter 11 Review

Chemical Reactions

Section 3: Predict the products of chemical reactions and balance. Indicate precipitates and aqueous solutions as known.

1. _____ Ag(s) + _____ $\text{CuSO}_4\text{(aq)}$ \rightarrow _____ + _____
2. _____ NaI(aq) + _____ $\text{CaCl}_2\text{(aq)}$ \rightarrow _____ + _____
3. _____ $\text{O}_2\text{(g)}$ + _____ $\text{H}_2\text{(g)}$ \rightarrow _____
4. _____ $\text{HNO}_3\text{(aq)}$ + _____ Mn(OH)_2 \rightarrow _____ + _____
5. _____ $\text{AgNO}_2\text{(aq)}$ + _____ $\text{BaSO}_4\text{(aq)}$ \rightarrow _____ + _____
6. _____ HCN + _____ $\text{CuSO}_4\text{(aq)}$ \rightarrow _____ + _____
7. _____ LiBr(aq) + _____ $\text{Co(SO}_3)_2$ \rightarrow _____ + _____
8. _____ HNO_3 + _____ Fe(OH)_3 \rightarrow _____ + _____
9. _____ C_5H_{12} + _____ $\text{O}_2\text{(g)}$ \rightarrow _____ + _____
10. _____ $\text{AgNO}_3\text{(aq)}$ + _____ Cu(s) \rightarrow _____ + _____

Identify the type of reaction for those equations above.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

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