

KINETIC THEORY OF MATTER WORKSHEET

1. A student performed a simple experiment using liquid bromine. She placed a small amount in the bottom of a sealed gas jar of air. After two minutes, red-brown fumes were seen just above the liquid bromine surface. After one hour, the red-brown fumes had spread completely throughout the gas jar.



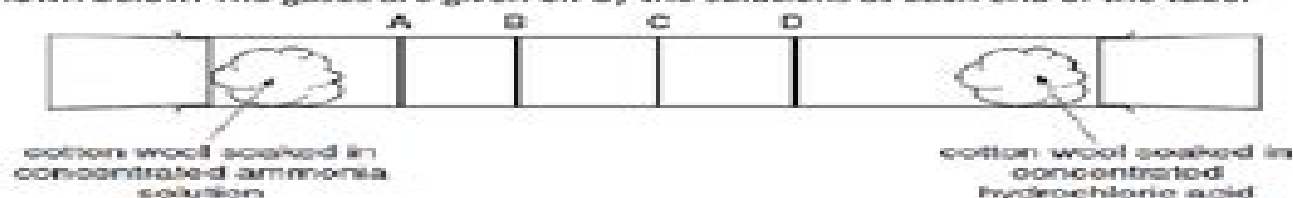
Explain these observations using the kinetic theory of matter

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2. The diffusion of ammonia gas and hydrogen chloride gas in a glass tube are shown below. The gases are given off by the solutions at each end of the tube.



- a. What is formed when ammonia gas mixes with hydrogen chloride gas?
-
- b. Which line shows where the product will be formed?
-
- c. Give a reason for your answer in b above using Graham's law of diffusion
-
- d. Is the product formed a solid, liquid or gas?
-

Kinetic Theory Section Reinforcement Answers

J Elliott



Kinetic Theory Section Reinforcement Answers:

The Enigmatic Realm of **Kinetic Theory Section Reinforcement Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Kinetic Theory Section Reinforcement Answers** a literary masterpiece penned by way of a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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