7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative positive crisscross noble gas electrons anion

octet ions cation electron configuration polyatomic empirical formula

- An ionic bond is an attraction between oppositely charged __ions
- 2. Anions have a __negative _____ charge
- 2. An atom becomes an ion by losing or gaining ____electrons
- The _octet __rule states that eterns tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
- When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) _noble gas
- The atoms that make up polyatomic __ions are bonded together by covalent bonds.
- The _crisscress _method can be used to write the formula for an ionic compound.
- 8. The _empirical formula if a compound denotes the ratio of loss in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

9. Calcium becomes a monatomic cotion by gaining two electrons.

true

10. A cation has a positive charge.

tirue

- A binary tonic compound contains only one kind of extion and one land of arrion.
- _polyatomic
- 12. Monatomic ions consist of more than one atom.

seven

- The Lewis dot diagram for chlorine, a group 7A element, has all electrons.
- true (released)
- A great deal of energy is <u>produced</u> when an ionic compound is formed.

Kinetic Theory Section 1 Reinforcement Answers

Anthea Maton

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