

## Analytical Chemistry MCQ Test. September'21.

1. The metal hydroxide which is insoluble in caustic soda but soluble in excess of ammonium hydroxide.
1.  $\text{Cu}(\text{OH})_2$                       2.  $\text{Zn}(\text{OH})_2$                       3.  $\text{Fe}(\text{OH})_2$                       4. None
2. The salt solution which does not react with ammonium hydroxide is -----
1. Calcium nitrate                      2. Zinc nitrate
3. Lead nitrate                      4. Copper nitrate
3. The metal oxide which can react with acids as well as alkali is-----.
1. Silver oxide                      2. Copper (II) oxide
3. Aluminum oxide                      4. Calcium oxide
4. Al metal shows amphoteric behaviour. it can be shown by
1. Its reaction with dil. NaOH only.
2. Its reaction with dil. HCl only.
3. Its reaction with both dil. HCl and dil. NaOH .
4. It does not react with either dil. HCl or dil. NaOH

### QUESTION 5

Which of the following hydroxides soluble in excess of NaOH?

1.  $\text{Cu}(\text{OH})_2$                       2.  $\text{Zn}(\text{OH})_2$                       3.  $\text{Fe}(\text{OH})_2$                       4.  $\text{Fe}(\text{OH})_3$
6. The salt which gives a dirty green precipitate with sodium hydroxide solution is.
1. Iron (III) sulphate                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate
7. The salt which gives a reddish-brown precipitate with sodium hydroxide solution is.
1. Iron (III) sulphate                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate
8. The salt which gives a gelatinous white precipitate with Ammonium hydroxide solution is.
1. Iron (III) chloride                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate
9. The salt which gives a chalky white precipitate with Ammonium hydroxide solution is.
1. Lead nitrate                      2. Iron (II) sulphate
3. Copper sulphate                      4. Zinc sulphate

# Mcq Pharmaceutical Analytical Chemistry

**J Rink**



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