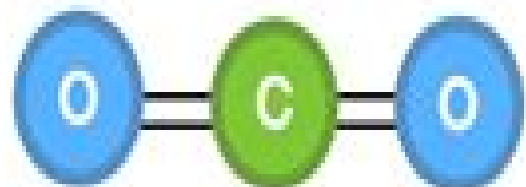
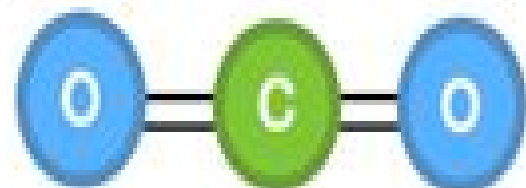


Visualizing Proust's Law of Definite Proportions



1 gram of CO₂ is:

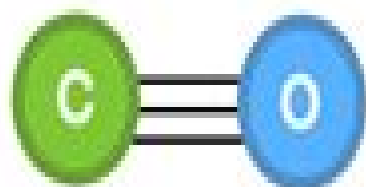
72.7% Oxygen (or 0.727g)
27.3% Carbon (or 0.273g)



16 grams of CO₂ is:

72.7% Oxygen (or 11.64g)
27.3% Carbon (or 4.364g)

The same ratio = the same compound



1 gram of CO is:

57.1% Oxygen (or 0.571g)
42.9% Carbon (or 0.429g)

Different ratios = different compounds

Law Of Definite Proportions Experiment Copper Sulfide

**Erwin Burr Kelsey, Harold Glenn
Dietrich**



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