

## 10 Honors Ionic and Metallic Bonding Questions [21 marks]

1. [1 mark]

Which is correct for all solid ionic compounds?

A. High volatility

B. Poor electrical conductivity

C. Low melting point

D. Good solubility in water

2. [1 mark]

Which combination corresponds to a strong metallic bond? B

|    | Charge on the metal ion | Radius of ion |
|----|-------------------------|---------------|
| A. | large                   | large         |
| B. | large                   | small         |
| C. | small                   | small         |
| D. | small                   | large         |

3. [1 mark]

How does a lithium atom form the most stable ion?

A. The atom gains a proton to form a positive ion.

B. The atom loses a proton to form a negative ion.

C. The atom loses an electron to form a positive ion.

D. The atom gains an electron to form a negative ion.

4. [1 mark]

What is the formula of ammonium phosphate?

# Ionic And Metallic Bonding Questions

**Rachel Sandford**



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