

Section Review

Objectives

1. Explain the formation of ionic compounds.
2. Describe the properties of ionic compounds.

Vocabulary

- | | |
|-------------------|-------------------|
| a. ionic compound | b. cation |
| c. anion | d. ionic compound |
| e. ionic crystal | |

Part A Completion

Write the word or words that best complete each statement. Write complete sentences that will demonstrate your understanding of each statement. Be specific.

1. Ions are charged atoms or molecules. An atom that has lost one or more electrons is called a(n) cation. An atom that has gained one or more electrons is called a(n) anion. Ions are formed by the loss or gain of electrons from a neutral atom.

2. A(n) ionic compound is a compound that is made up of ions. It is formed by the combination of a cation and an anion. The resulting compound is electrically neutral. The ions in the compound are held together by strong electrostatic forces called ionic bonds.

Part B True-False

Write True or False for each statement. Be specific if you are False. Explain why you are False.

1. True The formation of an ionic compound involves the transfer of electrons from one atom to another.

2. True Ions are charged particles.

3. True A cation is a positively charged ion.

4. True An anion is a negatively charged ion.

5. True Ionic compounds are made up of ions.

6. True The ions in an ionic compound are held together by ionic bonds.

7. True Ionic compounds are electrically neutral.

8. True The formation of an ionic compound involves the transfer of electrons.

9. True The resulting compound is electrically neutral.

10. True The ions in the compound are held together by strong electrostatic forces.

11. True The resulting compound is called an ionic compound.

12. True The resulting compound is called a salt.

Ionic Bonds Chapter 7 Section 7 2 Answers

K. Peter C. Vollhardt, Neil E. Schore



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