

## Mole and Stoichiometry Mini-Lab

Calculations: Show **ALL** work (bridge/grid).

The **GIVEN** for calculations #4 and #5 should be measurements, not results from previous calculations.

4. How many moles of water are in the graduated cylinder?  
UNKNOWN: mol  $\text{H}_2\text{O}$       GIVEN: g  $\text{H}_2\text{O}$  (data table)
5. How many molecules of water are in the graduated cylinder?  
UNKNOWN: molecules  $\text{H}_2\text{O}$   
GIVEN: g  $\text{H}_2\text{O}$  (data table)
6. For an experiment, 4.75 moles of water are needed.
  - a) How many grams of water are needed?
  - b) How many milliliters of water would this be?  
UNKNOWN: g  $\text{H}_2\text{O}$       GIVEN: 4.75 mol  $\text{H}_2\text{O}$   
UNKNOWN: mL  $\text{H}_2\text{O}$       GIVEN: 4.75 mol  $\text{H}_2\text{O}$

# Mini Lab 11 Apply Stoichiometry

**A Loxley**



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