

(ii) What would you observe during **reaction 1**?

.....
 [1]

(iii) In **reaction 1**, compound **E** was heated under reflux with excess $\text{K}_2\text{Cr}_2\text{O}_7/\text{H}_2\text{SO}_4$.

Suggest why these conditions were used, rather than the reaction mixture being distilled during the process.

.....

 [1]

(iv) Name the type of reaction taking place in **reaction 2**.

..... [1]

(b) Describe a chemical test that you could use to detect the presence of a carbonyl group in an organic compound.

Reagent

Observation

.....
 [2]

(c) Compound **E** is a single stereoisomer.

(i) Draw the skeletal formula of one other stereoisomer of compound **E** and state the type of stereoisomerism.

type of stereoisomerism [2]

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Yan Bai



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