

	Species	Lewis Structure	Molecular Geometry		Polar (answer yes or no and indicate polarity on Lewis drawing)
			Model drawing	VSEPR (electron geometry name and molecular structure name from Table 1)	
c.	BH_3	$\text{H} \cdots \text{B} \cdots \text{H}$ H	H $\text{H}-\text{B}-\text{H}$	Trigonal Planar	no
f.	NO_3^-	$\text{O} :: \text{N} :: \text{O}$ O	$\text{O}=\text{N}^--\text{O}$	Trigonal planar Trigonal planar	no
g.	HCO_2^-	$\text{H} \cdots \text{C} \cdots \text{O} :$ O	$\text{H}-\text{C}=\text{O}$ O^-	Trigonal planar	yes
h.	NF_3	$\text{F} \cdots \text{N} \cdots \text{F}$ F	$\text{F}-\text{N}-\text{F}$ F	Tetrahedral Pyramidal	yes
i.	CCl_4	$\text{Cl} \cdots \text{C} \cdots \text{Cl}$ Cl	$\text{Cl}-\text{C}-\text{Cl}$ Cl	Tetrahedral	yes
j.	H_3O^+	$\text{H} \cdots \text{O} : \text{H}$ H	$\text{H}-\text{O}^+-\text{H}$ H	Tetrahedral Pyramidal	yes
k.	H_2S	$\text{H} \cdots \text{S} \cdots \text{H}$ S	$\text{H}-\text{S}-\text{H}$	Tetrahedral Bent	yes
l.	OBr_2	$\text{Br} \cdots \text{O} \cdots \text{Br}$	$\text{Br}-\text{O}-\text{Br}$	Tetrahedral Bent	yes

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Lei Huang



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