

NATURAL SCIENCES

Gr. 9 - LESSON PLAN - MATTER & MATERIALS

TOPIC: COMPOUNDS		
Sub-topic: Names of Compounds		Duration: 1 Hour
CONTENT & CONCEPTS (CAPS p63)		
KEY CONCEPTS <ol style="list-style-type: none"> 1. Chemical reaction 2. Chemical formula 3. Names of compounds according to their elements. 		
Specific Aims:	Specific Aim 1: 'Doing Science'	X
	Specific Aim 2: 'Knowing the subject content and making connections'	X
	Specific Aim 3: 'Understanding the uses of Science'	
LESSON OBJECTIVES: <ol style="list-style-type: none"> 1. Learners will be able to write the names of compounds as indicated on page 63 of the CAPS Document. 2. Learners will be able to make models of different compounds 		
RESOURCES REQUIRED: <ul style="list-style-type: none"> • Periodic Table Poster • Siyavula Sasol Inzalo (Learner book) page 163 		
TEACHING & LEARNING ACTIVITY: Introduction <u>Briefly review and revise concepts dealt with in Grade 8, focusing on compounds</u> As each element has a unique name for example: <ul style="list-style-type: none"> - Hydrogen - Lithium Remember the first 20 elements learnt in grade 8. How did you memorize them? Do you remember Mnemonics... He - H; He - He; Lies - Li; Because - Be; Boys - B; Can - C; Not - N; Operate - O; Fireplaces - F; New - Ne; Nation - Na; Might - Mg; Also - Al; Sign - Si; Peace - P; Security - S; Clause - Cl A - Ar; King - K; Can - Ca LEARNERS MAY RECITE THIS MNEMONIC AS A WAY OF MEMORIZING THE FIRST 20 ELEMENTS. <u>The teacher provides the following explanation to the learners</u> Each compound has a unique name. How is a compound formed? <ul style="list-style-type: none"> - A compound is formed when TWO or more different elements combine chemically. - Each atom in a compound are held by chemical bonds - Compounds are formed during chemical reaction (when elements/ compounds combine to form new substances) - Each compound can be written as a formula, for example, H₂O; NaCl; CuSO₄ What is a formula ? <ul style="list-style-type: none"> - This is a notation that shows which elements the compound is made up of. - It indicates the total number of atoms in one unit of the compound - It indicates the ratio in which atoms of different elements are bonded to each other - It gives the total number of atoms in one unit of a compound. For example, the formula for Copper Sulphate is CuSO ₄		

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The Science Teacher, 2000 SCC Library has 1964 cur **Resources in Education**, 1986 **Rules and Regulations and Classification of Positions ... Adopted November 14, 1913** California. Civil Service Commission, 1914 Education at a Glance 2023 Sources, Methodologies and Technical Notes OECD, 2023-09-12 This publication provides additional reference material for Education at a Glance 2023 the authoritative source for data on the state of education around the world It provides guidance on the data and methodology used to calculate each indicator It also contains references to the sources and specific notes for each country The Natural Classroom Jack R. Edelman, 1996 A directory of field courses programs and expeditions in the natural sciences Bulletin, 1918 *Bulletin of the Buffalo Society of Natural Sciences* Buffalo Society of Natural Sciences, 1910 Vol 18 1938 Seventy five years a history of the Buffalo society of natural sciences 1861 1936 3 p 1 5 204 p *Recording for the Blind & Dyslexic, ... Catalog of Books*, 1996 **1977 National Science Foundation Authorization** United States. Congress. House. Committee on Science and Technology. Subcommittee on Science, Research, and Technology, 1976 Science John Michels (Journalist), 1913 Resources in Education, 1996 **Catalogue** University of North Dakota, 1934 **A Natural Science Program for the Intermediate Grades** Madison (Wis.). Board of Education, 1937 *A Natural Science Program for the Intermediate Grades: Grade V* Madison (Wis.). Board of Education, 1937 **A Natural Science Program for the Intermediate Grades: Grade VI** Madison (Wis.). Board of Education, 1937 American Secondary Education, 1975 *Current Index to Journals in Education*, 1997 **School Life**, 1957 **The Reality of Aid 1998-1999** Judith Randel, Tony German with Deborah Ewing, 2013-11-05 NOW IN ITS SIXTH ANNUAL EDITION The Reality of Aid has for the first time analysed the fair share of bilateral aid for basic social services basic education basic health reproductive health nutrition clean water and sanitation that should come from each donor an analysis which shows only two donors meeting their fair share and the G7 nations Canada France Germany Italy Japan UK US falling behind by over US 5 billion This year and next The Reality of Aid focuses on basic education as a right and not a privilege and its role in development cooperation and poverty elimination A key feature of The Reality of Aid 1998 1999 is the ten chapters offering analysis of development cooperation from the perspective of southern NGOs Many of these focus on basic education and raise issues around transparency gender and civil society If policies were programmes and promises were dollars The Reality of Aid could report great progress on the road to eradicating global poverty this year But at a time when donors acknowledge that ending poverty is possible it seems that commitments are being offered instead of resources and real change From the Summary Part I presents a useful Summary highlighting the steps that donors could take now to make progress towards poverty eradication and reviews the trends in development cooperation debt relief measures targeted towards the new millennium and commitment to the goal of ensuring basic education for all Part II gives a full report on the overseas aid performance of OECD country aid donors and the European Union over the last year Part III sets out a Southern

perspective on development cooperation Part IV provides at a glance comparisons of donors aid outlook and commitment to development cooperation in the 21st Century poverty eradication gender and public support Part V contains handy reference material Throughout the book information is presented in easily interpreted diagrams and graphs The Reality of Aid has established itself as a unique source of independent evaluation and comment on aid policies and development It is indispensable for all in the field whether in the official or voluntary sectors Indispensable it gives you most of the hard facts you need to know about the major issues New Internationalist The most comprehensive and rigorous independent analysis of the aid and development policies of the world s wealthiest nations Essential reading Charity World The Reality of Aid remains an essential purchase by the libraries of development institutions and an invaluable reference for development practitioners Development Change A reliable watchdog for anyone interested in this important aspect of international relations ORBIT An accessible reference it encompasses many key issues and stimulates further research Commonwealth Comparative Politics Originally published in 1998 **Membership & NSTA Publications Catalog** National Science Teachers Association,1997

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Table of Contents Natural Science Grade 9 Scope

1. Understanding the eBook Natural Science Grade 9 Scope
 - The Rise of Digital Reading Natural Science Grade 9 Scope
 - Advantages of eBooks Over Traditional Books
2. Identifying Natural Science Grade 9 Scope
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Natural Science Grade 9 Scope
 - User-Friendly Interface
4. Exploring eBook Recommendations from Natural Science Grade 9 Scope

- Personalized Recommendations
- Natural Science Grade 9 Scope User Reviews and Ratings
- Natural Science Grade 9 Scope and Bestseller Lists
- 5. Accessing Natural Science Grade 9 Scope Free and Paid eBooks
 - Natural Science Grade 9 Scope Public Domain eBooks
 - Natural Science Grade 9 Scope eBook Subscription Services
 - Natural Science Grade 9 Scope Budget-Friendly Options
- 6. Navigating Natural Science Grade 9 Scope eBook Formats
 - ePub, PDF, MOBI, and More
 - Natural Science Grade 9 Scope Compatibility with Devices
 - Natural Science Grade 9 Scope Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Natural Science Grade 9 Scope
 - Highlighting and Note-Taking Natural Science Grade 9 Scope
 - Interactive Elements Natural Science Grade 9 Scope
- 8. Staying Engaged with Natural Science Grade 9 Scope
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Natural Science Grade 9 Scope
- 9. Balancing eBooks and Physical Books Natural Science Grade 9 Scope
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Natural Science Grade 9 Scope
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Natural Science Grade 9 Scope
 - Setting Reading Goals Natural Science Grade 9 Scope
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Natural Science Grade 9 Scope

- Fact-Checking eBook Content of Natural Science Grade 9 Scope
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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