

Name : _____

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Multi-Step Equations: Integers

Level 1: S1

Solve each equation.

1) $3(x - 4) = 2(-2x + 1)$

2) $8q + 6 = 4q - 14$

3) $9 = \frac{v + 4}{v + 12}$

4) $7 - (5t - 13) = -25$

5) $-3(7p + 5) = 27$

6) $14 + 13y = 20y - 21$

7) $\frac{-8 - 3k}{2} = 11$

8) $-15b + 21 + 5b = -19$

Multi Step Integers

Brendan G. Carr

A red circular graphic with a gradient, appearing as a semi-circle or a stylized arrow pointing to the right, located on the right side of the light blue bar.

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web katie ashcraft 41 subscribers part two shows how to work through the formulas and complete the math that goes along with the atomic mass of candium activity sciencium

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web name lily fleishman background candium lab period 5 isotopes are atoms of the same element with the same atomic number number of protons but different masses due to different numbers of neutrons in the nuclei the atomic mass of an element is the weighted average of the masses of the different isotopes

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web answers the purpose of the lab summarizes observations clearly explains connection to calculating average atomic masses for actual elements answers the purpose of the lab summarizes observations clearly answers the purpose of the lab overall lab 12 points scaled to 95

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web if asked how you find an average most people respond that you add up the values and divide by how many there were the average of 2 2 4 and 8 is 16 4 or

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web procedure 1 obtain a sample of candium 2 separate the three isotopes and count obtain the total mass for each group of isotopes 3 complete the data table using the instructions found under the data analysis questions section the large x s in the table indicate cells that do not require data

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web candium is an interesting element found only at fordham high school for the arts we will be using this special new element today to learn about atomic mass calculations the atomic mass listed on the periodic table for each element is a weighted average

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web 1 obtain a sample of candium separate the 3 isotopes candium red hots mini candium peanut butter regular and candium peanut large and record the numbers for each type of isotope 2 next determine the mass for each isotope and calculate the total mass in grams 3 calculate the average mass

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web cole quinn period 5 mr hans honors chemistry candium post lab questions 1 no the average atomic mass of my group s candium sample was very similar to our neighbors samples if each group were to have used larger samples of candium the respective differences would have been smaller because the larger the sample size the greater the

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web obtain a sample of candium from your teacher 2 separate the three isotopes m m s skittle s and peanut m m s and measure the mass of each isotope record in data table 1 1 3 count the number of m m s skittle s and peanut m m s record in data table 1 1

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web procedure 1 place the candium atoms in a paper towel or a small bag 2 gently shake the atoms for 30 seconds this is the estimated half life of your candy 3 gently pour out the atoms onto a paper towel 4 count the number of atoms with the print side up these atoms have decayed 5

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web sample of candium do not eat electronic balance safety when handled correctly candium has no known safety hazards candium samples should always be placed in a paper cup for all weighings or stored on a piece of clean paper to avoid any contact with the laboratory bench surface or the balance pre lab questions write questions and

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web be able to calculate the atomic mass of a fictitious element called candium obtain the mass number for the isotopes of candium this lab activity includes 2 page lab activity step by step calculations organized data tables post lab questions step by step procedure teacher s prep guide post lab

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web 1 pt in the box to the right type your evaluation of how the recipe turned out i feel like it went well however it was a little hard to get out and i lost some crystals in the process 1 pt in the box to the right list chemical reaction evidence and or physical change s here