

CBSE PAPER-2024 (SET-1)
CHEMISTRY

- The molar ionic conductivities of Ca^{2+} and Cl^- are 119.0 and $76.3 \text{ S cm}^2 \text{ mol}^{-1}$ respectively. The value of limiting molar conductivity of CaCl_2 will be : [1]
 (a) $195.3 \text{ S cm}^2 \text{ mol}^{-1}$ (b) $43.3 \text{ S cm}^2 \text{ mol}^{-1}$
 (c) $314.3 \text{ S cm}^2 \text{ mol}^{-1}$ (d) $271.6 \text{ S cm}^2 \text{ mol}^{-1}$
- Consider the following reaction : [1]



Identify A and B from the given options :

- (a) A – Methanol, B – Potassium formate
 (b) A – Ethanol, B – Potassium formate
 (c) A – Methanol, B – Ethanol
 (d) A – Methanol, B – Potassium acetate
- Which of the following acids represents Vitamin C ? [1]
 (a) Saccharic acid (b) Gluconic acid
 (c) Ascorbic acid (d) Benzoic acid
- Rosenmund reduction is used for the preparation of Aldehydes. The catalyst used in this reaction is : [1]
 (a) $\text{Pd} - \text{BaSO}_4$ (b) Anhydrous AlCl_3
 (c) Iron (III) oxide (d) HgSO_4
- Which alkyl halide from the given options will undergo $\text{S}_{\text{N}}1$ reaction faster ? [1]
 (a) $(\text{CH}_3)_3\text{C}-\text{Br}$ (b) $(\text{CH}_3)_2\text{CH}-\text{Br}$ (c) $\text{CH}_3-\text{CH}_2-\text{Br}$ (d) $(\text{CH}_3)_3\text{C}-\text{CH}_2-\text{Br}$
- From the elements of 3d series given below, which element shows the maximum number of oxidation states ? [1]
 (a) Scandium (b) Manganese (c) Chromium (d) Titanium

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Daniel F McAuley



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