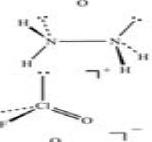
support of this structure.

1.

 S<sub>2</sub>O<sub>4</sub><sup>2-</sup> has SO<sub>2</sub> units with an angle of about 30° between their planes, in an eclipsed conformation.



 N<sub>2</sub>H<sub>4</sub> has a trigonal pyramidal shape at each N, and a gawche conformation. There is one lone pair on each N.



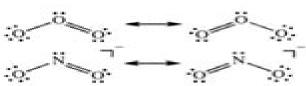
g. CIOF<sub>2</sub><sup>\*</sup> is a distorted trigonal pyramid with one lone pair and double bond character in the CI—O bond.



h. CS<sub>2</sub>, like CO<sub>2</sub>, is linear with double bonds. S=C=S

The structure of XeOF<sub>5</sub> is based on a pentagonal bipyramid, with a lone pair and the oxygen atom in axial positions. See K. O. Christe et al., *Inorg. Chem.*, **1995**, *34*, 1868 for evidence in

- 3.11 All the halate ions are trigonal pyramids; as the central atom increases in size, the bonding pairs are farther from the center, and the lone pair forces a smaller angle. The decreasing electronegativity Cl > Br > I of the central atom also allows the electrons to be pulled further out, reducing the bp-bp repulsion.
- 3.12 a. AsH<sub>3</sub> should have the smallest angle, since it has the largest central atom. This minimizes the bond pair—bond pair repulsions and allows a smaller angle. Arsenic is also the least electronegative central atom, allowing the electrons to be drawn out farther and lowering the repulsions further. Actual angles: AsH<sub>3</sub> = 91.8°, PH<sub>3</sub> = 93.8°, NH<sub>3</sub> = 106.6°.
  - b. CI is larger than F, and F is more electronegative and should pull the electrons farther from the S, so the F—S—F angle should be smaller in OSF<sub>2</sub>. This is consistent with the experimental data: the F—S—F angle in OSF<sub>2</sub> is 92.3° and the CI—S—CI angle in OSCI<sub>2</sub> is 96.2°.
  - c. NO<sub>2</sub><sup>-</sup> has rather variable angles (115° and 132°) in different salts. The sodium salt (115.4°) has a slightly smaller angle than O<sub>3</sub> (116.8°). The N-O electronegativity difference should pull electrons away from N, reducing the bp-bp repulsion and the angle.



- d. BrO<sub>3</sub> (104°) has a slightly smaller angle than ClO<sub>3</sub> (107°), since it has a larger central atom. In addition, the greater electronegativity of Cl holds the electrons closer and increases bp-bp repulsion.
- 3.13 a. N<sub>3</sub> is linear, with two double bonds. O<sub>3</sub> is bent (see solution to 3.12.c), with one double bond and a lone pair on the central O caused by the extra pair of electrons.

# **Miessler Tarr 4th Edition Solutions Manual**

John G. Verkade

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web diagram indicating kinetochore microtubules bound to kinetochores and the aster the aster is an array of microtubules that radiates out from the centrosome towards the cell edge diagram also indicates the centromere region of a chromosome the narrow waist where the two sister chromatids are most tightly connected and the kinetochore a

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web a1 external protection device activated malfunction of indoor unit pcb e5 e6 e6 e7 e9 e8 external protection device connected to the terminal strip k t1 t2 of indoor unit is activated e0 e1 unifi ed protection devices activated direct expansion commissioning vrv daikin internet - Oct 09 2022

web yes no vacuum test start pressure vacuum test time vacuum test end pressure visual inspection air side outdoor unit correct water side outdoor unit correct air side indoor unit correct piping distance  $\emptyset 64$  mm piping distance  $\emptyset 95$  mm piping distance  $\emptyset 127$  mm piping distance  $\emptyset 127$ 

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is true why have the units ran for the last 4 years i cant understand why both units are faulting with the same fault some suggestions on what to check would be good also this fault started with ja and e3 faults on all the controls but **troubleshooting of vrv system non wsq daikin** - Apr 15 2023

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