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arrhenius s definition of acids and bases the earliest definition of acids and bases is arrhenius s definition which states that an acid is a substance that forms hydrogen ions H^+ when dissolved in water and a base is a substance that forms hydroxide ions OH^- when dissolved in water

10 1 acids properties and examples chemistry libretexts - May 20 2023

mar 24 2021 acids react with bases to produce a salt compound and water when equal moles of an acid and a base are combined the acid is neutralized by the base the products of this reaction are an ionic compound which is labeled as a salt and water

5 3 acids bases and neutralization reactions - Sep 12 2022

the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is $\text{acid} + \text{base} \rightarrow \text{water} + \text{salt}$

14 2 acids properties and examples chemistry libretexts - Apr 19 2023

may 20 2018 acids react with bases to produce a salt compound and water when equal moles of an acid and a base are combined the acid is neutralized by the base the products of this reaction are an ionic compound which is labeled as a salt and water

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nov 13 2022 suggest simple tests you could carry out to determine if an unknown substance is an acid or a base state the chemical definitions of an acid and a base in terms of their behavior in water write the formula of the salt formed when a given acid and base are combined