

10 Honors Ionic and Metallic Bonding Questions [21 marks]

1. [1 mark]

Which is correct for all solid ionic compounds?

A. High volatility

B. Poor electrical conductivity

C. Low melting point

D. Good solubility in water

2. [1 mark]

Which combination corresponds to a strong metallic bond? **B**

	Charge on the metal ion	Radius of ion
A.	large	large
B.	large	small
C.	small	small
D.	small	large

3. [1 mark]

How does a lithium atom form the most stable ion?

A. The atom gains a proton to form a positive ion.

B. The atom loses a proton to form a negative ion.

C. The atom loses an electron to form a positive ion.

D. The atom gains an electron to form a negative ion.

4. [1 mark]

What is the formula of ammonium phosphate?

Ionic And Metallic Bonding Questions

Clifford Lane



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