

1. Electrolyte can conduct electricity, because \_\_\_\_\_  
(a) molecules have unpaired electrons  
(b) .molecules have paired electrons  
(c ).molecules have loosely held electrons  
(d) molecules have ions
2. Which one of the following is electrolyte?  
(a). benzene  
(c). sodium chloride  
(b). choloroform  
(d). Sugar
3. Which one of the following is not a weak electrolyte?  
(a).  $\text{NH}_4\text{OH}$   
(c).  $\text{CH}_3\text{COOH}$   
(b).  $\text{KOH}$   
(d).  $\text{H}_2\text{CO}_3$
4. Which of the following is not a strong electrolyte?  
(a).  $\text{NaCl}$   
(c).  $\text{KOH}$   
(b).  $\text{NaOH}$   
(d).  $\text{CH}_3\text{COOH}$
5. The electrode potential is the tendency of metal \_\_\_\_\_  
(a). to gain the electron  
(c) . either loss and gain of electron  
(b). To loose electron  
(d). None of the above
6. An electrochemical cell converts \_\_\_\_\_  
(a). electrical energy in chemical energy  
(b). chemical energy into electrical energy  
(c) chemical energy into heat energy  
(d). electrical energy into heat energy
7. Electrochemical cell is also known as \_\_\_\_\_  
(a). electrolytic cell  
(c). Daniel cell  
(b). galvanic cell  
(d). none of these
8. The passage of electricity in the Daniel cell , when zinc and copper electrodes are connected is from

# Mcqs About Electrochemistry

**Yicheng Fang**



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