

# Electron Energy and Light

How does light reveal the behavior of electrons in an atom?

## Why?

From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms. Trace materials, such as evidence from a crime scene, lead in paint or mercury in drinking water, can be identified by heating or burning the materials and examining the color(s) of light given off in the form of bright-line spectra.

## Model 1 – White Light



- Trace the arrows in Model 1 and shade in the table with colored pencils where appropriate.
- What happens to white light when it passes through a prism?

*The different wavelengths get separated*

- Why are the color labels in the table in Model 1 plural (i.e., "Reds" rather than "Red")?

*many different shades of a color in a range of wavelength*

- Do all colors of light travel at the same speed?

*Yes*

- Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively?

*All colors do not have the same energy. Blues + violets highest  
Reds + oranges lowest*

- Consider the light illustrated in Model 1.

- Which color corresponds to the longest wavelengths?

*Reds*

- Which color corresponds to the shortest wavelengths?

*Violets*

- Write a sentence that describes the relationship between wavelength and energy of light.

*As the wavelength gets shorter the energy of light increases*

# Light And Color Packet Answer

**Muhammad Maqbool**



## **Light And Color Packet Answer:**

*A Practical Guide to Stage Lighting* Steven Louis Shelley, 2013-05-02 The long awaited new edition of this celebrated bestseller     Normal Instructor and Primary Plans ,1918     Normal Instructor and Teachers World ,1918     An Introduction to Non-Ionizing Radiation Muhammad Maqbool, 2023-11-13 An Introduction to Non Ionizing Radiation provides a comprehensive understanding of non ionizing radiation NIR exploring its uses and potential risks The information is presented in a simple and concise way to facilitate easy understanding of relevant concepts and applications Chapters provide a summary and include relevant equations that explain NIR physics Other features of the book include colorful illustrations and detailed reference lists With a focus on safety and protection the book also explains how to mitigate the adverse effects of non ionizing radiation with the help of ANSI guidelines and regulations An Introduction to Non Ionizing Radiation comprises twelve chapters each explaining various aspects of non ionizing radiation including Fundamental concepts of non ionizing radiation including types and sources Interaction with matter Electromagnetic fields The electromagnetic wave spectrum UV visible light IR waves microwaves and radio waves Lasers Acoustic waves and ultrasound Regulations for non ionizing radiation Risk management of non ionizing radiation The book is intended as a primer on non ionizing radiation for a broad range of scholars and professionals in physics engineering and clinical medicine     **TCI** ,1996

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**Board Journal** William George Bruce, William Conrad Bruce, Susanna Cocroft, 1914 A periodical of school administration  
**The Living Church Annual and Clergy-list Quarterly**, 1888      **Philippics** Marcus Tullius Cicero, 1924 Introduction  
and notes in English parallel English and Latin text      *Home Garden and Flower Grower*, 1925      *American Druggists'*  
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