

LIFE SCIENCES - Grade 10**TOPIC TEST: History of Life on Earth****MARKS:30****TIME: 30 minutes**

SECTION A**QUESTION 1**

Various options are given as possible answers to the following questions. Choose the answer and write only the letter (**A to D**) next to the question number (**1.1 to 1.5**) in the **ANSWER BOOK**, for example **1.6 D**.

- 1.1 Palaeontology is a study of ...
- A the half-life of radioactive substances.
 - B continental drift.
 - C fossils.
 - D changes during the Cambrian explosion.
- 1.2 Which method allows scientists to say that one fossil is older than the other?
- A Fossilization
 - B Relative dating
 - C Petrification
 - D Radiometric dating
- 1.3 A period in the earth's history where global temperatures plummet and there is extensive glaciation is known as ...
- A the greenhouse effect.
 - B a cold spell.
 - C a hole in the ozone.
 - D an ice age.
- 1.4are the richest sources of fossils
- A Layers of strata
 - B Index fossils
 - C Mineral salts
 - D Sedimentary rocks

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determine based on electronegativity and symmetry if the following molecules are polar or nonpolar hcn h c n cf₄ c f₄ cocl₂
c o c l₂ nf₃ n f₃ q3

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molecule is one where the charge distribution around the molecule is not symmetric it results from having polar bonds and
also a molecular structure where the bond polarities do not cancel

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web when the atoms in a bond are not the same the bonding electrons are shared and the bond is electronegativities equally
unequally non polar polar unequally polar the degree of polarity of a bond between any two atoms is determined by
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bonds including covalent ionic and hydrogen bonds and london dispersion forces introduction living things are made up of
atoms but in most cases those atoms aren t just floating around individually

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molecules and ions contain polar bonds which of these molecules and ions have dipole moments a clf₅ b clo₂ clo₂ c tecl₂ 4
tecl₄ 2 d pcl₃ e sef₄ f ph₂ ph₂ g xef₂ answer problem 6 2 3 6 2 3

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the
molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry
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