

# Solubility Rules

## Mnemonic Tricks

Always Soluble  
NAG SAG

**N**itrates ( $\text{NO}_3^-$ )

**A**cetates ( $\text{C}_2\text{H}_3\text{O}_2^-$ )

**G**roup 1 ( $\text{Li}^+$ ,  $\text{Na}^+$ , etc.)

**S**ulfates ( $\text{SO}_4^{2-}$ )

**A**mmonium ( $\text{NH}_4^+$ )

**G**roup 17 ( $\text{F}^-$ ,  $\text{Cl}^-$ , etc.)

Exceptions  
PMS and Castro Bear

**P** ( $\text{Pb}^{2+}$ , lead)

**M** (mercury,  $\text{Hg}^{2+}$ )

**S** (silver,  $\text{Ag}^+$ )

**Ca**  $^{2+}$

**Sr**  $^{2+}$

**Ba**  $^{2+}$

# Model Rules Of Solubility In Aqueous Solutions Answers

**American Chemical Society**



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**Scientific American** ,1902      The National Druggist ,1900      *The Pharmaceutical Journal and Transactions* ,1878  
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**Chemistry in the Community.** American Chemical Society,2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts      Journal of the Chemical Society Chemical Society (Great Britain),1913      **Review of Physiological Chemistry** Harold Anthony Harper,1979      Chemistry William R. Robinson,Jerome D. Odom,Henry Fuller Holtzclaw,1992

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