## Identifying the 5 Types of Chemical Reactions

#### Identify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- · Double Replacement Reaction
- Combustion Reaction

#### Reactions (identify pattern below)

#### Name of Reaction Type

1. 
$$2 \underbrace{Na}_{A} + \underbrace{Cl_{2}}_{A} \Rightarrow 2 \underbrace{NaCl}_{AB}$$
 (A & B combine) Ext. synthesis

2.  $Pb + FeSO_{4} \Rightarrow PbSO_{4} + Fe$ 

3.  $P_{4} + 3 O_{2} \Rightarrow 2 P_{2}O_{3}$ 

4.  $2 NO_{2} \Rightarrow 2 O_{2} + N_{2}$ 

5.  $Na_{3}PO_{4} + 3 KOH \Rightarrow 3 NaOH + K_{3}PO_{4}$ 

6.  $C_{3}H_{6}O + 4 O_{2} \Rightarrow 3 CO_{2} + 3 H_{2}O$ 

7.  $MgCl_{2} + Li_{2}CO_{3} \Rightarrow MgCO_{3} + 2 LiCl$ 

8.  $C_{6}H_{12} + 9 O_{2} \Rightarrow 6 CO_{2} + 6 H_{2}O$ 

9.  $CaCO_{3} \Rightarrow CaO + CO_{2}$ 

10.  $2 AgNO_{3} + Cu \Rightarrow Cu(NO_{3})_{2} + 2 Ag$ 

11.  $2H_{2} + O_{2} \Rightarrow 2H_{2}O$ 

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**Terry C. Jones** 

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