

Test "Chemistry"

Chapter # 3

Objective part

Time: 15 min

Total marks: 12

Q.1- Every question has four possible answers, Encircle the correct option.

- i) The atomic radii of the elements in periodic table;
 - a) increase from left to right in a period
 - b) increase from top to bottom in a group
 - c) do not change from left to right in a period
 - d) decrease from top to bottom in a group
- ii) Mendeleev Periodic table was based upon the:
 - a) electronic configuration
 - b) atomic mass
 - c) atomic number
 - d) electro negativity
- iii) The amount of energy given out when an electron is added to an atom is called;
 - a) ionization energy
 - b) lattice energy
 - c) electron affinity
 - d) electro negativity
- iv) Long form of periodic table is constructed on the basis of:
 - a) Mendeleev postulate
 - b) atomic number
 - c) atomic mass
 - d) mass number
- v) Along the period which of the following decreases:
 - a) electro negativity
 - b) ionization energy
 - c) atomic radius
 - d) electron affinity
- vi) Mark the incorrect statement about the ionization energy:
 - a) it is measured kJ mol^{-1}
 - b) it is the absorption of energy
 - c) it decreases in a period
 - d) it decreases in groups
- vii) Transition elements are:
 - a) all gases
 - b) all metals
 - c) all non metals
 - d) all metalloids
- viii) S-f series of inner transition elements are called:
 - a) lanthanides
 - b) actinides
 - c) halogens
 - d) alkali metals
- ix) If we move from left to right in a period the value of ionization energy:
 - a) remain same
 - b) decreases
 - c) increases
 - d) not effected
- x) Which scientist has given the idea of actives in periodic table:
 - a) Mendeleev
 - b) Al-razi
 - c) Newland
 - d) dobernier
- xi) Shielding effect is greater in atoms with greater number of:
 - a) proton
 - b) electron
 - c) neutron
 - d) positron
- xii) Which is the incomplete period in the periodic table:
 - a) 4th period
 - b) 2nd period
 - c) 7th period
 - d) 1st period

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Frances Talaska Fischbach



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