

# Chemical Quantities Outline

## (The Molar Mass)

### Molar Mass

- Atomic Mass is the mass of a single atom expressed in amu; relative values based on the isotopes; (decimals) on the periodic table
- Molar mass is the mass of a mole of an element/compound/molecule; equal to the atomic mass  
Units are g/mole (grams per mole)
- Example:

Carbon's atomic mass 12.01 amu and its molar mass is 12.01 g/mol

### How to find the mass of a mole of a compound

- You must know the chemical formula
- Then add the atomic masses of the atoms that make up that compound
- Example: Sulfur Trioxide ( $\text{SO}_3$ )

1 Sulfur  $\rightarrow 32.06 \text{ g (1)} = 32.06 \text{ g/mol S}$   
 3 Oxygens  $\rightarrow 16.00 \text{ g (3)} = 48.00 \text{ g/mol O}$   
 $\rightarrow 80.06 \text{ g/mol SO}_3$

The decomposition of hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) provides sufficient energy to launch a rocket. What is the molar mass of hydrogen peroxide?

2 Hydrogen  
2 Oxygen

$2(1.008 \text{ g/mol}) \text{ H} = 2.016 \text{ g/mol H}$   
 $2(16.00 \text{ g/mol}) \text{ O} = 32.00 \text{ g/mol O}$   
 $= 34.02 \text{ g/mol H}_2\text{O}_2$

### Find the molar mass of calcium nitrate

1 Calcium  
2 Nitrogen  
6 Oxygen  
 $\text{Ca}(\text{NO}_3)_2$   
 $1(40.08 \text{ g/mol Ca}) + 2(14.01 \text{ g/mol N}) + 6(16.00 \text{ g/mol O})$   
 $40.08 \text{ g/mol} + 28.02 \text{ g/mol} + 96 \text{ g/mol} = 164.1 \text{ g/mol Ca}(\text{NO}_3)_2$

### Mole to Mass Relationship

- Molar mass of a substance is the mass in grams per mole of that substance
- Use the molar mass of an element or compound to convert between mass and the mole
- Conversion factors:  $\text{molar mass} = \frac{\text{g}}{\text{mol}}$  or  $\frac{1 \text{ mol}}{\text{g}}$

# Molar Mass And Chemical Quantities Answer Key

**Trace Jordan, Neville R. Kallenbach**



## **Molar Mass And Chemical Quantities Answer Key:**

**Chemistry for the Biosciences** Jonathan Crowe, Tony Bradshaw, 2014 Leading students through the essential concepts that are central to understanding biological systems this text uses everyday examples and analogies to build their confidence in an often daunting subject By focusing on the key themes that unify the subject it shows how integral chemistry is to the biosciences

**Chemistry** Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives

**Advances in Computer Assisted Learning** P. R. Smith, 2014-05-23 Advances in Computer Assisted Learning contains selected proceedings from the CAL Symposium on Computer Assisted Learning held at the University of Nottingham in the UK in 1985 This book reviews advances in computer assisted learning in the areas of curriculum development visually handicapped and disabled students project work in schools television viewdata and video applications database applications and engineering education and training This monograph has 35 chapters and opens with a discussion on the computing aspects of interactive video focusing on the design and production of the software used to control the videodisc developed by the Open University in the UK The next chapter illustrates a variety of case studies whereby local viewdata has been exploited by both teachers and their pupils in different parts of Europe Attention then turns to the use of computer assisted communication in the education of the visually impaired the use of microcomputers in teaching electronics and theoretical considerations in selecting software for language arts This text will be of interest to educators and policymakers who want to implement computer technology in the classroom

*Excel Revise in a Month TEE Chemistry* Roy Skinner, 2003

**Fundamentals of Chemistry** Ralph A. Burns, 1995 *Basic Chemistry* Steven S. Zumdahl, 1999 For a full description see catalog entry for Zumdahl Introductory Chemistry A Foundation 4 e

**Chemistry** Karen C. Timberlake, 1992

**Essentials of Chemistry** Ralph A. Burns, John William Hill, 1995 The second edition of this chemistry textbook that uses practice examples and applications relating chemistry to our lives and the environment

*Study Guide with Selected Solutions* Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

The Electrical Journal ,1878

**Prentice Hall Physical Science Concepts in**

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chemistry Dan Harris Quantitative Chemical Analysis provides a sound physical understanding of the principles of analytical chemistry and their applications in the disciplines      **Chemical News and Journal of Industrial Science** ,1905      **The Chemical News and Journal of Industrial Science** ,1919      **The Electrician** ,1878      *Chemistry* John McMurry,Robert C. Fay,2001 A colorful pedagogically enhanced standard textbook for the introductory course It begins with atomic structure proceeds next to bonding and molecules then to bulk physical properties of substances and ends with a study of chemical properties Each chapter concludes with a brief description of an interesting application or extension of the chapter subject a summary a list of key words and a large number of problems Many student oriented supplements are available Annotation copyright by Book News Inc Portland OR      **Prentice Hall Chemistry** ,2000      *Journal of Planar Chromatography--modern TLC* ,1991      *Handbook of Polyelectrolytes and Their Applications: Polyelectrolytes, their characterization and polyelectrolyte solutions* Sukant K. Tripathy,Jayant Kumar,Hari Singh Nalwa,2002      **CHEMISTRY** SILBERBERG,2003

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web it is usually made from ingredients including some or all of naphtha lanolin turpentine wax often called carnauba wax gum arabic ethylene glycol and if required a colourant such as carbon black or an azo dye such as aniline yellow wax comes from a substance between resins and fats

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web the embodiment of the invention provides a method for producing shoe polish by using wastewater generated in the production of gallic acid which comprises the following steps adding calcium hydroxide suspension into wastewater from gallic acid production mixing standing and filtering to obtain filtrate

*pdf formulation of a black leather shoe polish from hydrogenated* - Feb 10 2023

web mar 6 2005 the functional group of the prepared shoe polish was also identified using ftir analysis and the ftir peaks confirmed the presence of a long chain ester group in the shoe polish