

CHEMISTRY MOLE COOKIES

Practice
mole
calculations
with a
cookie
recipe!



Chemistry Mole Cookies Worksheet

Instructions to complete the recipe for chocolate chip cookies:

1. Fill a five table spoon butter into mixing bowl. (approximate molecular mass)
2. Add all the ingredients into the bowl with mixer. (approximate grams, then grams to the proper mixing units. Show the conversions below.)
3. Consider the recipe card on the back.

Note: Most of the ingredients are in baking, or food companies measure (forbids). They have been simplified here.

INGREDIENT	MOLECULAR MASS (molecular mass)	MOLES mole	GRAMS g	CONVERT UNIT amount?
flour $C_4H_8O_4$		1.00 moles		
baking soda $NaHCO_3$		0.0001 moles		
salt $NaCl$		0.004 moles		
brown $C_6H_{12}O_6$		1.000 moles		
white sugar $C_6H_{12}O_6$		0.004 moles		
brown sugar $C_6H_{12}O_6$		0.004 moles		
egg $C_4H_8O_4$		0.004 moles		
vanilla $C_8H_{10}O_2$		0.004 moles		
chocolate chips $C_6H_8O_2$		1.00 moles		

Use the following conversions to get from grams to moles (mixing units) of moles.

1 cup flour = 125.00 grams

1 cup brown sugar = 145.00 grams

1 large egg = 50 grams

1 teaspoon vanilla = 4.75 grams

1 ounce chocolate chips = 100.00 grams

1 cup white sugar = 100.00 grams

1 teaspoon baking soda = 0.004 grams

1 teaspoon salt = 0.004 grams

1 cup butter = 200.00 grams

continued on back

Mole Cookies Answers

Joacim Rocklöv



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