

STOICHIOMETRY: MASS – MOLE WORKSHEET

Solve the following stoichiometry problems and write the correct answer with units (For example: 34 mol ó 4.62 g) Round up to 2 decimals.

1. Based on the following chemical equation, please answer the following questions:



- a. Calculate the amount of moles of NaCl needed to produce 75g of FeCl₃.

Answer:

- b. Calculate the amount of Fe₂(SO₄)₃ in grams to produce 215 g of Na₂SO₄.

Answer:

- c. Calculate the amount in grams of FeCl₃ produced with 95g of Fe₂(SO₄)₃

Answer:

- d. Calculate the amount of moles of Fe₂(SO₄)₃ needed to react with 123.5 g of NaCl

Answer:

2. How many moles of CaCO₃ can be produced if we make 12.3g of Ca(HCO₃)₂ reacts with Ca(OH)₂?



Answer:

Mole To Mole Stoichiometry Problems Answers

AW Chickering



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

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The Practice of Chemistry Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink_POC_PowerPoint_Final.ppt 832KB

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