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POGIL- Stoichiometry

How do chemists use balanced chemical equations?

What happened to Avogadro when he got bit by 6.02 x 10 23 mosquitoes? He got Mol-aria

Why?

Chemists use balanced chemical equations as a basis to calculate how much reactant is needed or product is formed in a reaction. This is called <u>Stoichiometry-(stoi-key-ah-meh-tree)</u> Another way of looking at it is using the <u>mole ratio</u> from the balanced equation and information about one compound in the reaction to determine information about another compound in the equation. A <u>mole ratio</u> is a conversion factor <u>derived from the coefficients of a balanced chemical equation</u> interpreted in terms of moles. In chemical calculations, mole ratios are used to convert between moles one thing and moles of another.

Purpose: In this activity we will address the question: How do I convert between different chemical species in a given reaction?

MODEL 1: Mole-to-Mole Stoichiometry (1-step)

Example: mole-to-mole

Question: Given the following equation:

 $1 N_2(g) + 3 H_2(g) \rightarrow 2 NH_3(g)$

How many moles of H2 are needed to produce 34.8 moles of NH1?

Answer:

Mole ratio

34.8 mol NH₃ | 3 mol H₂ = 52.2mol H₂

2 mol NH₃

Key Questions:

- What was the given in the example above?
- 2. Where did the $\frac{3 \text{ mot} H_2}{2 \text{ mot} N H_3}$, come from?
- Double check to see if 52.2 moles of H2 is the correct answer. SHOW the math below.

Mole Ratios Pogil Answer

Nigel P. Freestone

Mole Ratios Pogil Answer:

Mole Ratios Lifelige, 2019 This lesson plan relates balanced chemical equations to everyday analogies defines stoichiometry and uses mole ratios to convert between amounts of substances in chemical reactions STOICHIOMETRY NARAYAN CHANGDER, 2024-04-01 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel https www youtube com smartguiziz I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging guiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today s academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests guizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams guizzes trivia and more

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