

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Pd: \_\_\_\_\_

### **MIXED** Stoichiometry Practice Review Problems

STEP 1: The following reaction will be used for this entire worksheet, please complete the balancing before going on.



STEP 2: Calculator the molar mass of each compound in the balanced reaction and write the values above the balanced reaction.

STEP 3: For aluminum sulfate, find the % of the metal in the compound. Please be sure to show your work. (Hint: WS.1 work)

**Al is the only metal, S and O are nonmetals. There are no metalloids in the formula.**

$$\% \text{ Al} = \frac{\text{total mass Al}}{\text{molar mass Al}_2(\text{SO}_4)_3} \times 100 \% = \frac{53.96 \text{ g}}{342.17 \text{ g}} \times 100 \% = 15.77 \%$$

STEP 4: Complete the following problems using **a factor chart with units & chemicals**. These problems could require ONE ratio, TWO ratios or even THREE ratios. Look for clues & use your mole map if necessary.

#### **PROBLEMS**

1) How many **grams of calcium sulfate** are produced from reacting **2.59 moles of aluminum sulfate**?

$$2.59 \text{ mol Al}_2(\text{SO}_4)_3 \times \frac{3 \text{ mol CaSO}_4}{1 \text{ mol Al}_2(\text{SO}_4)_3} \times \frac{136.15 \text{ g CaSO}_4}{1 \text{ mol CaSO}_4} = 1057.89 \text{ g CaSO}_4$$

moles A  $\longrightarrow$  moles B  $\longrightarrow$  grams B

# Modern Chemistry Chapter Mixed Review Stoichiometry Answers

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001      Basic Chemistry Steven S Zumdahl, 1996      Introductory Chemistry Darrell D. Ebbing, Albert E. Richardson, 1995      **Introductory Chemistry** Steven S. Zumdahl, 1996      Chemical Engineering Catalog, 1922      **Chemistry** John Olmsted, Gregory M. Williams, 1997 Textbook outlining concepts of molecular science      Quarterly Journal - Indian Chemical Society, 1974      **Opportunities in Protection Materials Science and Technology for Future Army Applications** National Research Council, Division on Engineering and Physical Sciences, Board on Army Science and Technology, National Materials Advisory Board, Committee on Opportunities in Protection Materials Science and Technology for Future Army Applications, 2011-08-27

Armor plays a significant role in the protection of warriors During the course of history the introduction of new materials and improvements in the materials already used to construct armor has led to better protection and a reduction in the weight of the armor But even with such advances in materials the weight of the armor required to manage threats of ever increasing destructive capability presents a huge challenge Opportunities in Protection Materials Science and Technology for Future Army Applications explores the current theoretical and experimental understanding of the key issues surrounding protection materials identifies the major challenges and technical gaps for developing the future generation of lightweight protection materials and recommends a path forward for their development It examines multiscale shockwave energy transfer mechanisms and experimental approaches for their characterization over short timescales as well as multiscale modeling techniques to predict mechanisms for dissipating energy The report also considers exemplary threats and design philosophy for the three key applications of armor systems 1 personnel protection including body armor and helmets 2 vehicle armor and 3 transparent armor Opportunities in Protection Materials Science and Technology for Future Army Applications recommends that the Department of Defense DoD establish a defense initiative for protection materials by design PMD with associated funding lines for basic and applied research The PMD initiative should include a combination of computational experimental and materials testing characterization and processing research conducted by government industry and academia

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