

CHAPTER 22 REVIEW***Organic Chemistry*****MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ A saturated organic compound
- (a) contains all single bonds.
 - (b) contains at least one double or triple bond.
 - (c) contains only carbon and hydrogen atoms.
 - (d) is quite soluble in water.
2. Arrange the following in order of increasing boiling point:
- _____ a. ethane
 - _____ b. pentane
 - _____ c. heptadecane
3. Recall that isomers in organic chemistry have identical molecular formulas but different structures and IUPAC names.
- _____ a. Two isomers must have the same molar mass. True or False?
 - _____ b. Two isomers must have the same boiling point. True or False?
4. Explain why hydrocarbons with only single bonds cannot form geometric isomers.

5. Write the IUPAC name for the following structural formulas:



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ML Morrison



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