

## Chemistry Chapter 5 Exam

### Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to
  - a. Mendeleev.
  - b. Moseley.
  - c. Bohr.
  - d. Ramsay.
- \_\_\_\_\_ 2. Mendeleev left spaces in his periodic table and predicted several elements and their
  - a. atomic numbers.
  - b. colors.
  - c. properties.
  - d. radioactivity.
- \_\_\_\_\_ 3. Mendeleev attempted to organize the chemical elements based on their
  - a. symbols.
  - b. properties.
  - c. atomic numbers.
  - d. electron configurations.
- \_\_\_\_\_ 4. Mendeleev noticed that properties of elements usually repeated at regular intervals when the elements were arranged in order of increasing
  - a. atomic number.
  - b. density.
  - c. reactivity.
  - d. atomic mass.
- \_\_\_\_\_ 5. Mendeleev is credited with developing the first successful
  - a. periodic table.
  - b. method for determining atomic number.
  - c. test for radioactivity.
  - d. use of X rays.
- \_\_\_\_\_ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar
  - a. properties.
  - b. atomic numbers.
  - c. densities.
  - d. colors.
- \_\_\_\_\_ 7. In developing his periodic table, Mendeleev listed on cards each element's name, atomic mass, and
  - a. atomic number.
  - b. electron configuration.
  - c. isotopes.
  - d. properties.
- \_\_\_\_\_ 8. Mendeleev predicted that the spaces in his periodic table represented
  - a. isotopes.
  - b. radioactive elements.
  - c. permanent gaps.
  - d. undiscovered elements.
- \_\_\_\_\_ 9. Mendeleev's table was called periodic because the properties of the elements
  - a. showed no pattern.
  - b. occurred at repeated intervals called periods.
  - c. occurred at regular time intervals called periods.
  - d. were identical.
- \_\_\_\_\_ 10. The person whose work led to a periodic table based on increasing atomic number was
  - a. Moseley.
  - b. Mendeleev.
  - c. Rutherford.
  - d. Cannizzaro.
- \_\_\_\_\_ 11. Moseley's work led to the realization that elements with similar properties occurred at regular intervals when the elements were arranged in order of increasing
  - a. atomic mass.
  - b. density.
  - c. radioactivity.
  - d. atomic number.
- \_\_\_\_\_ 12. Who used his experimental evidence to determine the order of the elements according to atomic number?
  - a. Meyer
  - c. Stas

# Modern Chemistry Chapter 5 Mixed Review Answers

**John O'M. Bockris, Amulya K.N. Reddy**



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**draw a well labelled diagram showing the alimentary canal of cockroach** - Aug 11 2023

web solution verified by toppr the development of periplaneta americana cockroach is of paurometabolous type meaning there is development through the nymphal stage the nymph looks very much similar to the adult form the nymph molts about 13 times before reaching the adult form