

# Molar Mass

# Worksheet-Answer Key

Name	Date

Exercise 1. How much does one mole weigh (molar mass/molecular weight) for the following? Show work: which numbers are added/multiplied, how many decimal places are used from the periodic table, and your result.

Na	23g	NaF	42g
	22.99 → 23 g		23 + 19 = 42  g
Zn	65.4 g _	СО	28 g
	65.38 → 65.4 g		12 + 16 = 28  g
solid iron	55.9 g _	$CO_2$	44 g
	Fe 55.85 → 55.9 g	12	$+2 \times 16 = 12 + 32 = 44 \text{ g}$
Ca <sup>2+</sup> (aq)	40 g	$CH_4$	16 g
	same as Ca 40.08 → 40 g		$12 + 4 \times 1 = 12 + 4 = 16$
CaCO <sub>3</sub>	100 g _	$C_4H_{10}$	58 g
	$40 + 12 + 3 \times 16 = 100g$		$4 \times 12 + 10 \times 1 = 58g$

# **Molar Mass Of Butane Worksheet Answers**

**J Spring** 

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