

Molarity Practice Worksheet

Find the molarity of the following solutions: Molarity =
mole/Liters

1 liter = 1000 mL

- 1) 2.5 moles of NaCl is dissolved to make 0.050 liters of solution.

- 2) 3.0 moles of AgCl is dissolved in enough water to make a 2.0 liter solution

- 3) 4.2 moles of lithium fluoride are dissolved in 2 liters of solution.

- 4) 18 grams of hydrochloric acid, HCl is dissolved to make 1 liter of solution.

Molarity And Molality Practice Worksheet Answers

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