



The Mole Study Guide

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Avogadro's number refers to the number of particles in one gram of a substance.

- ☐ True
- ☐ False

What is the mass of one mole of aluminum?

- ☐ 26.982 g
- ☐ 13 g
- ☐ 53.965 g
- ☐ 14 g

Which is the correct formula for the compound whose percent composition is shown?

Mole Study Guide Answers

Roman Wölfel

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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What is the change in atomic number after the alpha decay? It decreases by 2. b. ... answer the following questions.

Nuclear ... 25.2 Nuclear Transformations | Lecture notes Chemistry These nuclei decay by turning a neutron into a proton to emit a beta particle (an electron) from the nucleus. This process is known as beta emission. It ... 60 s - 1 min

SECTION 25.2 NUCLEAR TRANSFORMATIONS. 1. Write a nuclear equation for the following radioactive processes. a. alpha decay of francium-208 $^{208}\text{Fr} \rightarrow \text{b} \dots$

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