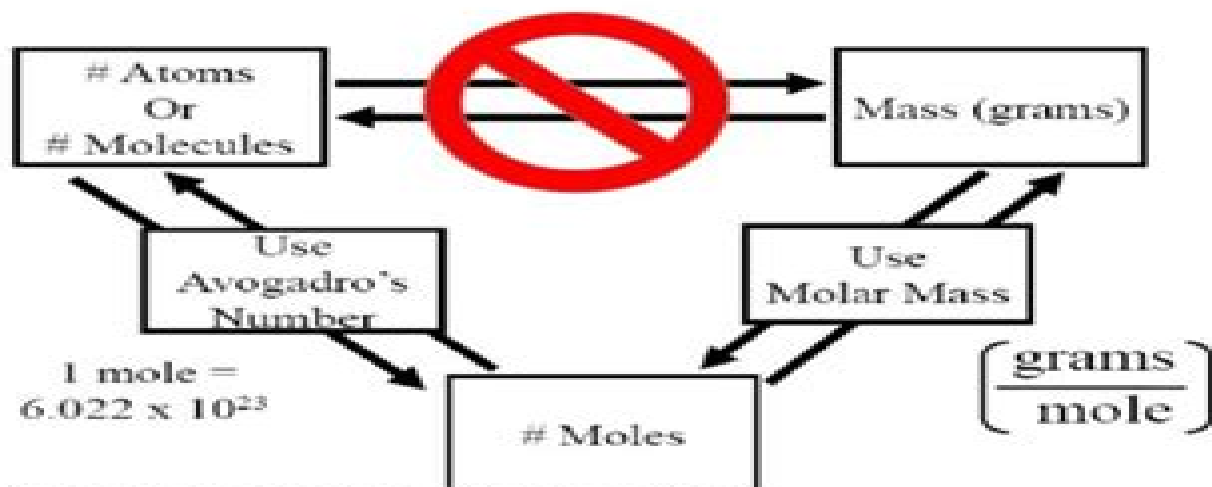


# Moles, Molecules, and Grams Worksheet and Key



1) How many moles are there in 24.0 grams of  $\text{FeF}_3$ ?

$$24 \div 19(3) + 56 = 0.212 \text{ mol}$$

2) How many moles are there in 458 grams of  $\text{Na}_2\text{SO}_4$ ?

$$458 \div 23(2) + 32 + 16(4) = 3.23 \text{ mol}$$

3) How many grams are there in  $2.30 \times 10^{24}$  atoms of silver?

$$2.30 \times 10^{24} \div 6 \times 10^{23} = 3.83 \text{ mol}$$

$$3.83 = y \div 108$$

$$y = 414\text{g}$$

4) How many grams are there in 7.40 moles of  $\text{AgNO}_3$ ?

$$7.40 = y \div 108 + 14 + 16 \times 3$$

$$7.40 = y \div 170$$

$$y = 1258\text{g}$$

# Moles Molecules And Grams Answers With Work

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