

## Colorimetry Pre-Lab Questions

Sec \_\_\_\_\_

Name \_\_\_\_\_

1. Consider the glassware found in the lab drawers at JCCC.

What glassware (be specific on the size) do you intend to use to hold (contain) the five solutions you will prepare? \_\_\_\_\_

Explain why you chose this glassware: \_\_\_\_\_

What glassware (be specific on the size) will you use to measure the volumes of the stock  $\text{CuSO}_4(\text{aq})$  and the water? \_\_\_\_\_

Explain why you chose this glassware: \_\_\_\_\_

2. Complete the table below and show a sample calculation for the volumes needed to prepare solution 2. Note: the total volume of each solution will be 10.0 mL and the stock solution has a  $\text{CuSO}_4$  concentration of 0.40 M.

Solution	$[\text{CuSO}_4]$	volume of 0.40 M $\text{CuSO}_4$	volume of $\text{H}_2\text{O}$
1	0.080 M		
2	0.16 M		
3	0.24 M		
4	0.32 M		
5	0.40 M		

Sample calculation for solution 2 (including units):

3. Graph the following data (x-axis = concentration, y-axis = absorbance). Label the axes, title the graph and staple it to this page.

Concentration (M)	0.000	0.020	0.040	0.060	0.080	0.100
Absorbance	0.000	0.165	0.329	0.484	0.688	0.873

*Note: You may use graph paper to plot the data by hand or you may use computer software to make your graph.*

# Jccc Chemistry 125 Lab Solution Manual

**David Kirk**



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