

केवल अभ्यास हेतु नमूना प्रश्न पत्र
Sample Question Paper for Practice only
हाईस्कूल परीक्षा - 2024
High School Examination-2024
विषय - हिन्दी
Subject - Hindi

Total Question	Total Printed Pages	Time	Maximum Marks
23	4	3 hours	75

MP Board Class 10th Hindi Model Paper 2024

- iv. प्रश्न क्रमांक 6 से 17 तक कुल 12 प्रश्न हैं। प्रत्येक प्रश्न पर 2 अंक निर्धारित हैं। शब्द सीमा लगभग 30 शब्द हैं।
v. प्रश्न क्रमांक 18 से 20 तक कुल 3 प्रश्न हैं। प्रत्येक प्रश्न पर 3 अंक निर्धारित हैं। शब्द सीमा लगभग 75 शब्द हैं।
vi. प्रश्न क्रमांक 21 से 23 तक कुल 3 प्रश्न हैं। प्रत्येक प्रश्न पर 4 अंक निर्धारित हैं। शब्द सीमा लगभग 120 शब्द हैं।
vii. प्रश्न क्रमांक 6 से 23 तक सभी प्रश्नों के आंतरिक विकल्प दिए गए हैं।

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JR Anderson



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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be
22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k
for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of
gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the
standard molar volume figure 9 18

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used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this
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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between

temperature and pressure the pressure of the 125 ml flask was at 101.36 kPa when the temperature was at 23.2 °C

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kPa temperature °C temperature K constant k p t 95.82 kPa 16.0 °C 289 K 0.33 97.90 kPa 19.0 °C 292 K 0.34 98.97 kPa 99.22 kPa 101.4 kPa 19.1 °C

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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