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web in a calorimeter the white solid $CaSO_4$ forms and the temperature of the mixture increases to 28.1 °C assuming that the calorimeter absorbs only a negligible quantity of heat and that the specific heat capacity of the solution is 4.18 J/g·°C and that the density of the final solution is 1.0 g/mL calculate the enthalpy change per mole

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web problem 8 5 1 6 8 5 1 6 when 50.0 g of 0.200 M NaCl(aq) at 24.1 °C is added to 100.0 g of 0.100 M AgNO₃(aq) at 24.1 °C in a calorimeter the temperature increases to 25.2 °C as AgCl(s) forms assuming the specific heat of the solution and products is 4.20 J/g °C calculate the approximate amount of heat in joules produced answer