

Calculating limiting reactants.

Let's look at an example:



In this reaction, 56 g nitrogen gas (N_2) was added to a chamber containing 14 g hydrogen gas (H_2). Which was the limiting reactant? (A_r : $\text{H}=1$, $\text{N}=14$)

First steps?

Limiting Reagent Exam Stoichiometry

**John Presti, George Wetzel, James
Colaizzi**



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Teaching General Chemistry Mansoor Niaz, 2008 The main objective of this monograph is to incorporate history and philosophy of science in the chemistry curriculum in order to provide students an overview of the dynamics of scientific research which involves controversies conflicts and rivalries among scientists that is the humanising aspects of science A major thesis of this book is the parallel between the construction of knowledge by the students and the scientists In looking for this relationship it is not necessary that ontogeny recapitulate phylogeny but rather to establish that students can face similar difficulties in conceptualising problems as those faced by the scientists in the past Given the vast amount of literature on students alternative conceptions misconceptions in science it is plausible to suggest that these can be considered not as mistakes but rather as tentative models leading to greater conceptual understanding Just as scientists resist changes in the

hard core of their beliefs by offering auxiliary hypotheses students may adopt similar strategies Conceptual change in science education can thus be conceptualised as building of tentative models that provide greater explanatory power to students understanding **Journal of the Pennsylvania Academy of Science** ,2001 *The Best Test Preparation & Review*

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