

LIMITING REACTANTS

1. What mass of CS_2 is produced when 17.5 g of C are reacted with 39.5 g of SO_2 according to the equation below?



What mass of the excess reactant will be left over?

2. What mass of NO is produced when 87.0 g of Cu are reacted with 225 g of HNO_3 according to the equation?



What mass of the excess reactant will be left over?

3. What mass of P_4 is produced when 41.5 g of $\text{Ca}_3(\text{PO}_4)_2$, 26.5 g of SiO_2 and 7.80 g of C are reacted according to the equation



How many grams of each excess reactant will remain unreacted?

Limiting Reactants And Percent Yield 9 3 Section Review

Jessica J Manson



Limiting Reagents And Percent Yield 9 3 Section Review:

Uncover the mysteries within Explore with is enigmatic creation, Embark on a Mystery with **Limiting Reaactants And Percent Yeild 9 3 Section Review** . This downloadable ebook, shrouded in suspense, is available in a PDF format (*). Dive into a world of uncertainty and anticipation. Download now to unravel the secrets hidden within the pages.

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