



Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Limiting Reactants

**Vocabulary:** chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

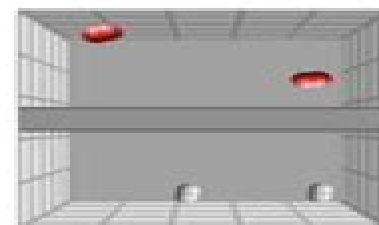
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make? \_\_\_\_\_ you can make 30 hotdogs with single bun and wieners \_\_\_\_\_
2. Which ingredient limited the number of hot dogs you could make? the buns having 30 limited the number of hot dog i can make \_\_\_\_\_
3. Which ingredient will you have leftovers of? i would have leftover of the wieners remaining 2 more. \_\_\_\_\_

### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.



To begin, make sure  $\text{H}_2 + \text{O}_2$  becomes  $\text{H}_2\text{O}$ . The small "2" in  $\text{H}_2$ ,  $\text{O}_2$ , and  $\text{H}_2\text{O}$  is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of  **$\text{O}_2$  molecules** and  **$\text{H}_2$  molecules** to two.
  - A. How many hydrogen molecules ( $\text{H}_2$ ) are there? \_\_\_\_\_ 1 \_\_\_\_\_ Hydrogen atoms? \_\_\_\_\_ 2 \_\_\_\_\_
  - B. How many oxygen molecules ( $\text{O}_2$ ) are there? \_\_\_\_\_ 1 \_\_\_\_\_ Oxygen atoms? \_\_\_\_\_ 2 \_\_\_\_\_
2. How many  $\text{H}_2\text{O}$  molecules do you think will form when these four molecules react? \_\_\_\_\_ 4 \_\_\_\_\_



# Limiting Reactants Gizmo Answers

**Lifelique**



**Limiting Reactants Gizmo Answers:**

Limiting Reactant and Percent Lifeliqe,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

Immerse yourself in the artistry of words with Experience Art with is expressive creation, **Limiting Reactants Gizmo Answers** . This ebook, presented in a PDF format ( Download in PDF: \*), is a masterpiece that goes beyond conventional storytelling. Indulge your senses in prose, poetry, and knowledge. Download now to let the beauty of literature and artistry envelop your mind in a unique and expressive way.

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