



LIMITING REAGENT AND PERCENT YIELD

Section Review

Objectives

1. Calculate limiting reagents (reagent in a reaction which determines the amount of product) (product not calculated) (12.3 objectives)
2. Calculate theoretical and actual yield, or percent and percent discrepancy (12.3 objectives)

Vocabulary

- | | |
|----------------------|------------------|
| 1. limiting reagent | 2. actual yield |
| 3. theoretical yield | 4. percent yield |
| 5. percent yield | |

Key Equations

1. percent yield = $\frac{\text{actual yield}}{\text{theoretical yield}} \times 100$

Part A Completion

Write the appropriate number for each space (use knowledge of the concepts, and knowledge of the correct number). Write what unit has been requested (where a unit is asked for) (where no unit is asked for)

1. Nitrogen gas (N₂) and hydrogen gas (H₂) react to form ammonia (NH₃). How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____
2. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____
3. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____
4. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____
5. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____
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10. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____

11. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____
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19. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____
20. How many grams of ammonia (NH₃) can be formed from 10.0 g of N₂ and 10.0 g of H₂? _____

Part B True-False

Write T for true and F for false (use knowledge of the concepts, and knowledge of the correct number)

1. The limiting reagent is the reagent that is completely used up in a reaction. _____
2. The limiting reagent is the reagent that is completely used up in a reaction. _____
3. The limiting reagent is the reagent that is completely used up in a reaction. _____
4. The limiting reagent is the reagent that is completely used up in a reaction. _____
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9. The limiting reagent is the reagent that is completely used up in a reaction. _____
10. The limiting reagent is the reagent that is completely used up in a reaction. _____

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Morris Hein, Susan Arena



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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity

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Chemistry 2007 David Wilson, 2006-12 Kaplan s guide includes 2 full length practice tests Diagnostic test to target areas for score improvement Detailed answer explanations Hundreds of practice questions from calculations of chemical equations to organic chemistry Explanations of important terms formulas and concepts Powerful strategies to help you score higher

Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and

biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Prentice Hall Chemistry*, 2000

Practice Makes Perfect Chemistry Review and Workbook, Second Edition Marian DeWane, Heather

Hattori, 2018-12-28 The Winning Equation for Success in Chemistry is Practice Practice Practice This book will help you apply concepts and see how chemistry topics are interconnected Inside are numerous lessons to help you better understand the subject These lessons are accompanied by dozens of exercises to practice what you've learned along with a complete answer key to check your work Throughout this book you will learn the terms to help you understand chemistry and you will expand your knowledge of the subject through hundreds of sample questions and their solutions With the lessons in this book you will find it easier than ever to grasp chemistry concepts And with a variety of exercises for practice you will gain confidence using your growing chemistry skills in your classwork and on exams YOU LL BE ON YOUR WAY TO MASTERING THESE

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Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application *Holt Chemistry* Ralph Thomas Myers, 2004 Chemical Principles with Qualitative Analysis William L. Masterton, Emil J. Slowinski, Conrad L. Stanitski, 1986 *Chemistry, Student Solutions Manual* James E. Brady, Frederick A. Senese, 2004-02-09 Work more effectively and check solutions as you go along with the

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Understanding Chemistry Fred M. Dewey, 1994 **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24

McGraw Hill Chemistry Review and Workbook Mary Millhollon, Richard H. Langley, 2022-09-23 The ideal tool for sharpening your chemistry skills This review guide and workbook will give you everything you need to excel in your high

school classwork and on standardized tests Clear and concise explanations walk you step by step through each essential chemistry concept 500 practical review questions in turn provide extensive opportunities for you to practice what you ve learned If you are looking for material based on national or state standards this book is your ideal study tool Features Designed to help you excel in the classroom and on standardized tests Concise clear explanations offer step by step instruction so you can easily grasp key concepts 500 review questions provide extensive opportunities for you to practice what you ve learned Aligned to national standards including the Common Core State Standards as well as the standards of non Common Core states and Canada

Introductory Chemistry Victor S. Krimsley, Darold E. Skerritt, Beverly B. Harrison, 1986 This pedagogically rich text has all the necessary features to hook introductory students and keep them interested and successful in preparatory chemistry This book carefully guides beginning students through the fundamental principals and calculations required for their subsequent success in either the general chemistry course or the short organic and biochemistry course Krimsley while dedicated to conceptual understanding and skill building presents a solid book that provides students with complete explanations on every point and helps them work through the material methodically with many examples and hints His approach is cohesive and coherent always patient and interesting Krimsley introduces all topics through an example students are already familiar with He continually reminds them of objectives and provides many opportunities to practice and then check their answers The text begins with a study of atomic and molecular structure prior to treating the various categories of chemical reactions The organization is designed to get students into chemistry quickly yet methodically The classification of matter begins in Chapter 2 before the chapter on measurement and the coverage of bonding appears in Chapter 8 The elementary concepts of chemistry are presented with an emphasis on mathematical calculations and problem solving strategies

Chemistry Oxtoby, 1994 **General Chemistry** James E. Brady, Gerard E. Humiston, 1982 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems

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