

The Chemical Earth

1. The living and non-living components of the Earth contain mixtures

1.2. Identify the difference between elements, compounds and mixtures in terms of particle theory.

Particle Theory

-All matter is made of tiny particles which are continually moving

Elements

-Elements are the basic building blocks of matter

-They are pure substances

-Cannot be broken down into simpler materials by chemical methods

-E.g. Iron

Compounds

-A compound is formed when two or more elements are chemically combined

-Compounds are pure substances

-Made up of two or more elements/different particles

-Can be broken down into constituent elements that make it up

-Properties different from those elements that make it up

-E.g. Aluminium oxide, water, sodium chloride, sodium carbonate

Mixtures

-Mixtures are most materials in the Earth

-They are impure substances

-Contains two or more types of particles which can be pure substance

-Has variable composition/proportions

-Can be separated by physical means (such as boiling and filtering)

-Has the properties of the pure substances it is composed of

→ Can be heterogenous or homogenous

→ Heterogenous: having non-uniform composition throughout so the appearance of the solid is not consistent (e.g. wood)

→ Homogenous: has uniform composition throughout so the appearance of the mixture is consistent (e.g. toothpaste)

-E.g. sea water, coffee, milk, petrol

Kiss Prelim Chemistry Topic Chemical Earth Answers

Michael Rajnik



Kiss Prelim Chemistry Topic Chemical Earth Answers:

Chemical Engineering Catalog ,1924 **Nuclear Science Abstracts** ,1975-10

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