

## THE SPECTROSCOPIC IDENTIFICATION OF ORGANIC FUNCTIONAL GROUPS

### IR and NMR Spectroscopy

- (1) The following is part of an IR spectrum (approximate reading):
  - a) Absorption bands (approximate values):
  - b) Assigning the compounds to name the compound(s) identified.
  - c) A molecular formula.
  - d) The name of the compound.
- (2) A compound with molecular formula  $C_8H_{10}O$  has IR spectrum (approximate reading) as follows:
  - a) Strongest bands of absorption (approximate):
  - b) Strongest bands of absorption (approximate):
  - c) The intensity of peaks.
  - d) The peak positions (cm<sup>-1</sup>).
- (3) Which one of the following IR bands is equivalent to the band in the IR spectrum of the compound?
  - a) 1715 cm<sup>-1</sup>
  - b) 1735 cm<sup>-1</sup>
  - c) 1755 cm<sup>-1</sup>
  - d) 1775 cm<sup>-1</sup>
- (4) The IR spectrum of a compound is shown below. The compound is:
  - a) 1,2-dichloroethane
  - b) 1,1-dichloroethane
  - c) 1,1,2,2-tetrachloroethane
  - d) 1,1,1,2-tetrachloroethane
- (5) A compound with molecular formula  $C_8H_{10}O$  has IR spectrum (approximate reading) as follows:
  - a) The compound is 1,2-dichloroethane and the compound is 1,1-dichloroethane.
  - b) The compound is 1,1,2,2-tetrachloroethane and the compound is 1,1,1,2-tetrachloroethane.
  - c) The compound is 1,1,2,2-tetrachloroethane and the compound is 1,1,1,2-tetrachloroethane.

# Kmlttb 2015 May Exam Date

**G Thomas**



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## **Table of Contents Kmlttb 2015 May Exam Date**

1. Understanding the eBook Kmlttb 2015 May Exam Date
  - The Rise of Digital Reading Kmlttb 2015 May Exam Date
  - Advantages of eBooks Over Traditional Books
2. Identifying Kmlttb 2015 May Exam Date
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Kmlttb 2015 May Exam Date
  - User-Friendly Interface
4. Exploring eBook Recommendations from Kmlttb 2015 May Exam Date
  - Personalized Recommendations
  - Kmlttb 2015 May Exam Date User Reviews and Ratings
  - Kmlttb 2015 May Exam Date and Bestseller Lists
5. Accessing Kmlttb 2015 May Exam Date Free and Paid eBooks
  - Kmlttb 2015 May Exam Date Public Domain eBooks
  - Kmlttb 2015 May Exam Date eBook Subscription Services
  - Kmlttb 2015 May Exam Date Budget-Friendly Options
6. Navigating Kmlttb 2015 May Exam Date eBook Formats

- ePub, PDF, MOBI, and More
- Kmlttb 2015 May Exam Date Compatibility with Devices
- Kmlttb 2015 May Exam Date Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Kmlttb 2015 May Exam Date
  - Highlighting and Note-Taking Kmlttb 2015 May Exam Date
  - Interactive Elements Kmlttb 2015 May Exam Date
- 8. Staying Engaged with Kmlttb 2015 May Exam Date
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Kmlttb 2015 May Exam Date
- 9. Balancing eBooks and Physical Books Kmlttb 2015 May Exam Date
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Kmlttb 2015 May Exam Date
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Kmlttb 2015 May Exam Date
  - Setting Reading Goals Kmlttb 2015 May Exam Date
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Kmlttb 2015 May Exam Date
  - Fact-Checking eBook Content of Kmlttb 2015 May Exam Date
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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