

## C1.1 Pre-Unit Quiz: Particles

### Mark Scheme

Qu	Answer	Marks	Supporting information for fix-it tasks
1	B	1	<p>Answering A shows a confusion with the equipment used to measure temperature and the actual units of temperature.</p> <p>Answering C shows a gap in knowledge that the unit of temperature includes 'Celsius'.</p> <p><i>Suggested fix-it: Write a sentence that includes how to measure temperature and what unit is used.</i></p>
2	A	1	<p>Answering B or C shows a misunderstanding of the 3 states of matter.</p> <p><i>Suggested fix-it: Write a sentence listing the three states of matter and give an example for each.</i></p>
3	A	1	<p>Answering B shows a confusion with solids and metals when grouping the objects. <i>Suggested fix-it: Write a sentence listing the three states of matter and give an example for each.</i></p> <p>Answering C shows some understanding of the properties of solids suggesting a confusion with the term 'state of matter'. <i>Suggested fix-it: Copy and complete this (gaps are underlined): 'There are three states of matter, <u>solid</u>, liquid and <u>gas</u>. Metals are a type of <u>material</u> that are usually <u>solid</u> at room temperature.'</i></p>
4	B	1	<p>Answering A or C shows a misunderstanding of the states of matter and classifying substances. Fizzy drinks are liquid with air bubbles (gas) within them.</p> <p><i>Suggested fix-it: List which states of matter are present in a tray of ice cubes melting.</i></p>
5	A	1	<p>Answering B or C shows a confusion with the properties of solids and liquids/gases. <i>Suggested fix-it: Compare how solids have different properties to liquids.</i></p>
6	C	1	<p>Answering A shows a misconception that gases form a pool (they actually flow to fill their container rather than form a pool like a liquid). <i>Suggested fix-it: Write a list of the ways that gases and liquids have similar properties.</i></p>

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid  $\text{H}_2\text{CO}_3$  and its conjugate base the bicarbonate ion  $\text{HCO}_3^-$

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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