

# International Kangaroo Mathematics Contest 2010

Benjamin Level: Class (5 & 6)

Max Time: 2 Hours

## 3-point problems

**Q1)** Knowing that  $\blacktriangle + \blacktriangle + 6 = \blacktriangle + \blacktriangle + \blacktriangle + \blacktriangle$ , which number is hidden by  $\blacktriangle$ ?

- A) 2      B) 3      C) 4      D) 6

**Q2)** The number 4 is next to two mirrors so it reflects twice as shown. When the same thing happens to number 5, what do we get instead for the question mark?



- A) 2      B) 2      C) 5      D) 5

**Q3)** Kalim goes directly from Zoo to School. He counts each flower on the way. Which of the following number can not be his result.

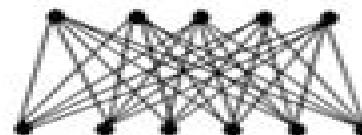


- A) 9      B) 10      C) 11      D) 12

**Q4)** A ladder has 21 stairs. Nadeem and Mahmood are counting stairs; one – from bottom to top, another – from top to bottom. They met on a stair that was called the 10<sup>th</sup> by Nadeem. What number will Mahmood give to this stair?

- A) 13      B) 11      C) 12      D) 10

**Q5)** Adil has connected all the upper points to all the lower points. How many lines Adil has drawn?

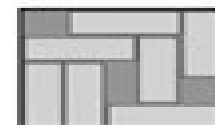


- A) 20      B) 25      C) 30      D) 35

**Q6)** A fly has 6 legs, while a spider has 8 legs. Together, 2 flies and 3 spiders have as many legs as 10 birds and

- A) 3 cats      B) 4 cats      C) 5 cats      D) 6 cats

**Q7)** There are seven bars in the box. It is possible to slide the bars in the box so there will be space for one more bar. At least how many bars have to be moved?



- A) 1      B) 2      C) 3      D) 4

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**Karin Nielsen-Saines**



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web lanthanides and actinides are series of elements with atomic numbers from 57 to 71 and from 89 to 193 respectively because their f electron shells are gradually filled as the atomic number increases 4f for lanthanides and 5f for actinides 1 these elements are known as f block elements figure 1

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third group of the periodic table the series is the row below the lanthanide series which is located underneath the main body of the periodic table lanthanide and actinide series are both referred to as rare earth metals these

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web the actinides are elements 89 to 103 and fill their 5f sublevel progressively actinides are typical metals and have properties of both the d block and the f block elements but they are also radioactive lanthanides have different chemistry from transition metals because their 4f orbitals are shielded from the atom's environment

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make up the f block of the periodic table the lanthanides are the elements produced as the 4f sublevel is filled with electrons and the actinides are formed while filling the 5f sublevel generally speaking the lanthanides

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web jun 27 2019 the lanthanides and actinides are arranged below with a notation in the main body cell rather than making room for all these elements in the 3b group in the main body of the periodic table because of the f shell electrons these two element groups are also known as the f block elements