Association of American Medical Colleges - Medical College Admission Test - AAMC MCAT Practice Exam 2

C/P: What expression gives the amount of light energy (in J per photon) that is converted to other forms between the fluorescence excitation and emission events?

"intensity of fluorescence emission at 440 nm excitation at 360 nm) was monitored for 20 minutes"

A) $(6.62 \times 10{\text -}34) \times (3.0 \times 108)$

B) $(6.62 \times 10{\text -}34) \times (3.0 \times 108) \times (360 \times 10{\text -}9)$

C) $(6.62 \times 10-34) \times (3.0 \times 108) \times [1/(360 \times 10-9) - 1/(440 \times 10-9)]$

D) (6.62 × 10-34) × (3.0 × 108) / (440 × 10-9) - **√√**C) (6.62 × 10-34) × (3.0 × 108) × [1 / (360 × 10-9) - 1 / (440 × 10-9)]

The answer to this question is C because the equation of interest is $E = hf = hc/\lambda$, where $h = 6.62 \times 10 - 34 \text{ J} \cdot \text{s}$ and $c = 3 \times 10 \text{ 8 m/s}$. Excitation occurs at $\lambda e = 360 \text{ nm}$, but fluorescence is observed at $\lambda f = 440 \text{ nm}$. This implies that an energy of $E = (6.62 \times 10 - 34) \times (3 \times 10 \text{ 8}) \times [1 / (360 \times 10 - 9) - 1 / (440 \times 10 - 9)]$ J per photon is converted to other forms between the excitation and fluorescence events.

C/P: Compared to the concentration of the proteasome, the concentration of the substrate is larger by what factor?

"purified rabbit proteasome (2 nM) was incubated in the presence of porphyrin...the reaction was initiated by addition of the peptide (100 uM)"

A) 5×101

B) 5×102

C) 5×103

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