Principle.		

COVALENT BOND WEBQUEST

Section.

DRECTIONS: Answer the following by going to the notes website. You can go to Mrs. Showlatt's website and use the links or type in the addresses for each section if needed.

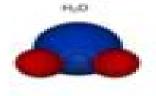
Activity #1- Introduction to Covalent Bonding.

Open Chemical Bonding Scroll down to the heading "Covalent Bonding." OR (http://www.visconiesrning.com/library/module_viewer.prp?mst-55)

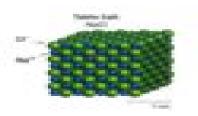
1. As opposed to		bonding in which a -	complete transfer of electrons	
occur.	bonding occurs when two or more elements			
	electrons, Gavale	and bioinding occurs beca	ause the atoms in the compound	
have a similar tendency	y for electrons (pr	enenally to	electrons). This	
most commonly occurs when two		bon	d together. Because both of the	
nonmetals will want to		electrons, the elements involved will share		
electrons in an effort to		their valence shells.		
3. How many valence s	nlectrons does hy	drogen need to have a	full first shell?	
4. How does the hydro		1050 		
5. What compound doe	ss hydrogen form	7		
6. How do hydrogen as				

8. Label the substances below as "lonic" or "covalent".

 Visit the simulation. Covatent bonding between hydrogen atoms. OR http://www.visionleaming.com/library/flash_viewer.php?coi+13466mid+55



Describe or draw what you see.



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Rachel Turoscy

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