

Chap 3 Test Practice

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Temperature is a measure of the average _____ energy of the particles in the object.

a. thermal	c. potential
b. kinetic	d. chemical
- _____ 2. The heavier a particle, the _____ it moves.

a. slower	c. less
b. faster	d. more
- _____ 3. What determines the speed of the atoms and molecules of a particular substance?

a. size of the atoms and molecules	c. Both (a) and (b)
b. temperature of the substance	d. None of the above
- _____ 4. Which state of matter will hold its shape without a container?

a. solid	c. gas
b. liquid	d. plasma
- _____ 5. The kinetic theory states that the higher the temperature, the faster the _____

a. particles that make up a substance move.
b. bonds between atoms break down.
c. molecules of gas rush together.
d. lighter particles within a substance clump together.
- _____ 6. The change of a substance from a solid directly to a gas is called _____

a. condensation.	c. melting.
b. evaporation.	d. sublimation.
- _____ 7. Evaporation refers to the change of state from a _____

a. liquid to a gas.	c. solid to a liquid.
b. gas to a liquid.	d. liquid to a solid.
- _____ 8. The law of conservation of mass states that mass cannot be _____

a. burned.	c. created or destroyed.
b. changed in form.	d. heated or cooled.
- _____ 9. Ice floats in water because it is _____

a. more dense than water.	c. colder than water.
b. less dense than water.	d. warmer than water.
- _____ 10. Buoyant force is the _____ force exerted on an object immersed or floating on a liquid.

a. lateral	c. downward
b. upward	d. tensile
- _____ 11. As the temperature of a fixed amount of gas at constant volume decreases, its pressure _____

a. decreases.	c. increases.
b. stays the same.	d. None of the above

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